

This article was downloaded by:

On: 29 January 2011

Access details: *Access Details: Free Access*

Publisher *Taylor & Francis*

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



## Phosphorus, Sulfur, and Silicon and the Related Elements

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t713618290>

## Recent Developments in the Chemistry of Cations and Anions Containing Low Valent Phosphorus Centers

E. Niecke<sup>a</sup>; G. David<sup>a</sup>; R. Detsch<sup>a</sup>; B. Kramer<sup>a</sup>; M. Nieger<sup>a</sup>; P. Wenderoth<sup>a</sup>

<sup>a</sup> Anorganisch-Chemisches Institut der Universität Bonn, Bonn 1, Germany

**To cite this Article** Niecke, E. , David, G. , Detsch, R. , Kramer, B. , Nieger, M. and Wenderoth, P.(1993) 'Recent Developments in the Chemistry of Cations and Anions Containing Low Valent Phosphorus Centers', *Phosphorus, Sulfur, and Silicon and the Related Elements*, 76: 1, 25 – 28

**To link to this Article:** DOI: 10.1080/10426509308032349

**URL:** <http://dx.doi.org/10.1080/10426509308032349>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.informaworld.com/terms-and-conditions-of-access.pdf>

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

## RECENT DEVELOPMENTS IN THE CHEMISTRY OF CATIONS AND ANIONS CONTAINING LOW VALENT PHOSPHORUS CENTERS

E. NIECKE\*, G. DAVID, R. DETSCH, B. KRAMER, M. NIEGER AND P. WENDEROTH

Anorganisch-Chemisches Institut der Universität Bonn,  
 Gerhard-Domagk-Str. 1, D-5300 Bonn 1, Germany

### Iminophosphenium- and Methylene-phosphenium-Cations

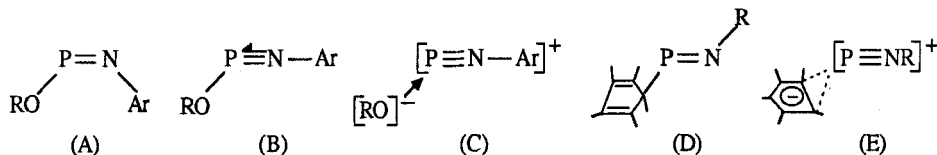
We have already demonstrated that the iminophosphenium ion,  $[P \equiv NAr]^+$  (Ar=2,4,6-*t*Bu<sub>3</sub>C<sub>6</sub>H<sub>2</sub>) **1** is produced by reaction of chloro(aryl)iminophosphine in the presence of the Lewis acid AlCl<sub>3</sub> [1].

In order to get further insight into the formation of this unique cationic system we studied the structural effects in iminophosphines produced by  $\sigma$ -acceptor substituents at phosphorus and  $\sigma$ -donor substituents at nitrogen, respectively.

Within a series of the oxo-derivatives, RO-P=N-Ar **2**, which has been obtained by nucleophilic displacement reactions from ClP=NAr, we have found that the electronic nature of the substituent R (Me<sub>3</sub>Si(**2a**), *t*Bu<sub>2</sub>CH(**2b**), *p*-Me-C<sub>6</sub>H<sub>4</sub>-SO<sub>2</sub>(**2c**), CF<sub>3</sub>SO<sub>2</sub>(**2d**)) exert not only a deformation of the geometry (A)  $\rightarrow$  (B) (P-N-C(Ar) 144.4(2) (**2a**); 179.1(4) (**2b**)) but also a P-O-bond elongation (B)  $\rightarrow$  (C) (P-O 162.8(3) (**2b**), 172.8(3) (**2c**), 192.3(3) (**2d**)).

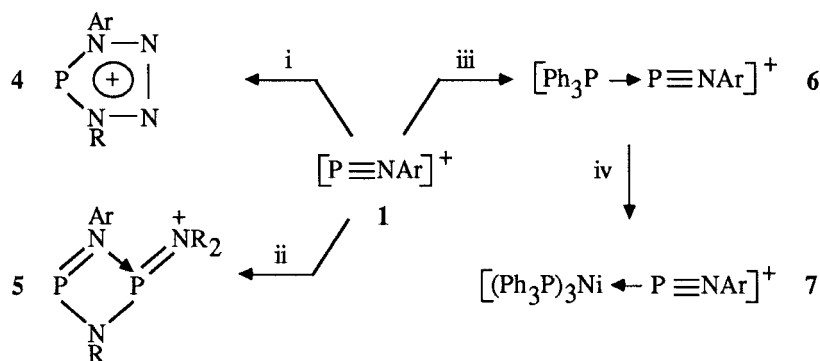
These effects (A)  $\rightarrow$  (C) are accompanied by a continuous PN-bond strengthening as well as a shielding of the <sup>31</sup>P nuclei and result in **2d** (P-N 146.7(3), P-N-C(Ar) 176.4(3),  $\delta^{31}P$  55) in a bonding situation similar to that found in **1**.

Similar effects are observed for iminophosphines of the type C<sub>5</sub>Me<sub>5</sub>-P=NR, **3**, with increasing the  $\sigma$ -donor ability of the nitrogen substituent. While a classical structure (D) with a P/C-bond order of unity (188.7(2)) has been observed in the case of R=(Me<sub>3</sub>Si)<sub>2</sub>N, a strong  $\sigma$ -donating substituent (R=*i*Pr<sub>3</sub>Si) results in a  $\eta^2$ -attachment of the planar Me<sub>5</sub>C<sub>5</sub>-ring to the phosphorus atom (P-C 212.2(4), 216.8(4) [2]), so that this species can be regarded as a  $\pi$ -complex between a cyclopentadienyl anion and an aza-phosphenium cation (E).

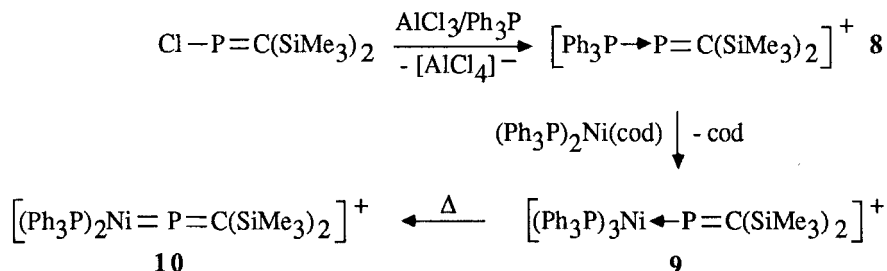


Reaction of the cation **1** with azides (i) or amino(imino)phosphines (ii) produces the corresponding [2+3]- **4** and [2+2]-cycloadducts **5**, the structures of which were studied by X-ray diffractometry. The averaged bond distances for **4**, are in accord with a delocalized  $6\pi$ -system, whilst the significant differences in the PN-distances in **5** (e.g. P-N(Ar) 163.6(3), 180.1(3);

P-N(R) 163.2(3), 179.0(3)) are in accord with the formation of an intramolecular donor/acceptor complex one mesomeric structure of which is displayed in **5**. Reaction of **1** with the donor  $\text{Ph}_3\text{P}$  (iii) gives the 1:1 adduct **6** ( $\delta^{31}\text{P}$  22, 87,  $J_{\text{PP}} = 340$  Hz), which by treatment with  $(\text{Ph}_3\text{P})_2\text{Ni}(\text{cod})$  (iv) is transformed into the azaphosphenium complex **7** via ligand shift from the phosphorus to the metal atom ( $\delta^{31}\text{P}$  28, 210,  $J_{\text{PP}} = 69$  Hz).



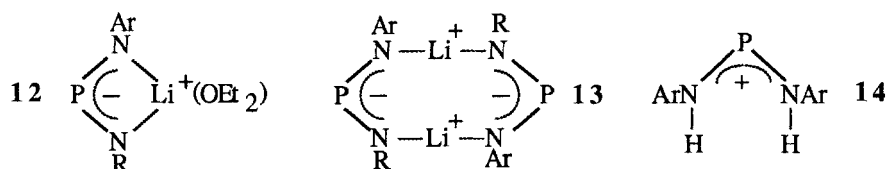
With respect to our investigations on **1** we studied also the interaction of chloro(methylene)phosphines with  $\text{AlCl}_3$  in the presence of suitable donor molecules and obtained by this route the stable phosphine-adduct **8**. It possesses a strong P/C (163.5(4)) and a weak P/P-bond (226.7(4)) and can be used as a suitable precursor for the methylene-phosphenium cation, as it is demonstrated by the reaction **8** with  $(\text{Ph}_3\text{P})_2\text{Ni}(\text{cod})$  to give the 18- and 16- electron nickel complexes, **9** ( $\delta^{31}\text{P}$  26, 504,  $J_{\text{PP}} = 77$  Hz) and **10** ( $\delta^{31}\text{P}$  29, 309,  $J_{\text{PP}} = 128$  Hz), respectively.



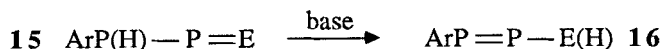
### 1,3-Diazaphosphaallyl-, 1-Azadiphosphaallyl-, 1,2-Diphosphaallyl- and 1-Phosphaallyl-Anions

Starting from NH-functionalized amino-iminophosphines,  $\text{RN(H)-P=NAr}^{[3]}$ , **11** ( $\text{R=tBu, 1-Ad, Mes, CPh}_3, \text{Ar}$ ), the corresponding 1,3-diaza-phosphaallylic systems were obtained by reaction with *n*-butyllithium. The lithium complexes exhibit a monomeric, **12** or dimeric structure, **13**, depending on the donor properties of the solvent and the steric demand of the substituents<sup>[3]</sup>.

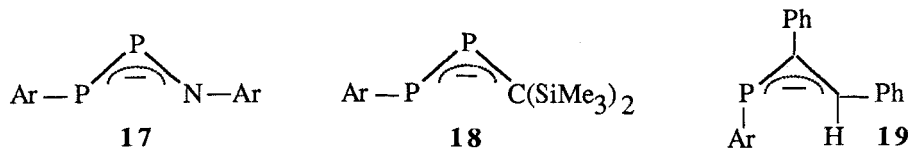
According to the X-ray structure, the bonding situation in the anion is comparable to the bis(amino)phosphenium cation, **14**  $\cdot [\text{AlCl}_4]^-$  (P-N 161.1, N-P-N 103.7).



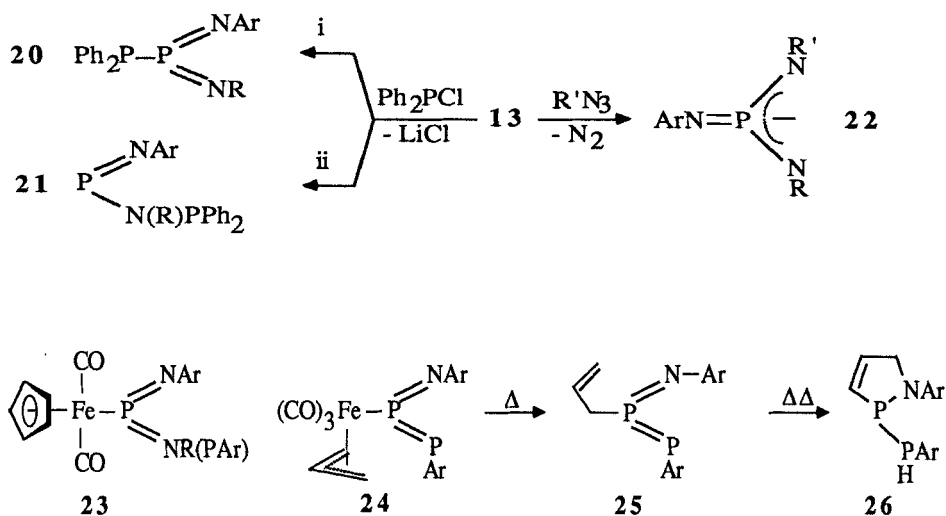
PH-functionalized phosphino(iminophosphines ( $\text{E=NAr}$ ) and -methylenephosphines ( $\text{E=C(SiMe}_3)_2$ ), **15** undergo base catalyzed isomerization into the corresponding diphosphene isomers **16**.



However, treatment of **15** with *n*-BuLi results in the formation of the lithium salts of 1-azadiphosphaallyl- **17** ( $\delta^{31}\text{P}$  244, 494,  $J_{\text{PP}}$  = 544 Hz) and 1,2-diphosphaallyl- anion **18** ( $\delta^{31}\text{P}$  212, 471,  $J_{\text{PP}}$  = 534 Hz). Furthermore, the synthesis of a stable 1-azaphosphaallyl system **19** was realized by treatment of  $\text{ArP(H)Li}$  with toluene. The heteroallylic structure of **17** and **19** is also proven by X-ray crystallographic studies (P-P 206.6(2), P-N 163.2(3), P-P-N 106(1),  $\text{N}\cdots\text{Li}$  213.9(7) (**17**); P-C 175.7(6), C-C 136.6(6), P-C-C 128.7(6),  $\text{P}\cdots\text{Li}$  > 300) (**19**).



Reactions of these novel heteroallyl anions include the ambident reaction behaviour of **13** towards  $\text{Ph}_2\text{PCl}$  to give the isomers **20** or **21** depending on the solvent used (i = ether, ii = pentane)<sup>[3]</sup>. Reaction with alkyl azide produces the tris(imino)metaphosphate anion **22**, whose allylic bonding situation is evidenced by the  $^{31}\text{P}$ - and  $^{15}\text{N}$ -n.m.r. data (e.g.  $\text{R}=\text{tBu}$ :  $\delta^{31}\text{P}$  64.9,  $\delta^{15}\text{N}$  - 234 (NAr), - 281 (NR),  $J_{\text{NP}} = 69$  Hz (PNAr), 30 Hz (PNR)). The ferrio-derivatives **23** and **24**, were obtained by the reaction of **13**(**17**) with  $(\eta^5\text{-C}_5\text{H}_5)\text{Fe}(\text{CO})_2\text{Cl}$  and **17** with  $(\eta^3\text{-C}_3\text{H}_5)\text{Fe}(\text{CO})_3\text{Cl}$  and it is shown that the latter (**24**) is easily transformed into the diphosphene-imine **25** and its phosphino-azaphospholene isomer **26**.



## ACKNOWLEDGEMENT

The authors thank the Deutsche Forschungsgemeinschaft (SFB 334) and the Fonds der Chemischen Industrie for generous financial support.

## REFERENCES

1. E. Niecke, M. Nieger, F. Reichert, *Angew. Chem. Int. Ed. Engl.* **1988**, *37*, 1715. 3.
2. D. Gudat, H. M. Schiffner, M. Nieger, D. Stahlke, A.J. Blake, H. Grondy, E. Niecke, *J. Am. Chem. Soc.*, in press.
3. R. Detsch, E. Niecke, M. Nieger, W.W. Schoeller, *Chem. Ber.* **1992**, *125*, 1119.